

tence of EHD was given by two experiments proving their macroscopic character, namely the photovoltaic detection⁴ of EHD in p-n junctions and the scattering⁵ of 3.39 μ light by EHD.

In the first experiment, a Ge p-n junction photodiode is immersed in pumped liquid helium at 2 K, and the n region is illuminated with a CW mercury lamp for instance. The new emission lines of Ge are easily seen to be present in these conditions of temperature and excitation. The fluctuations in the photovoltaic current are then analyzed with a rapid oscilloscope and are seen to have a spiky structure similar to that of the current from a photomultiplier. But at higher temperature (~ 10 K), this structure is not observed. Each current spike is interpreted as being due to the destruction in the junction electric field of an EHD containing N electron-hole pairs. By integrating the current under a spike, one easily gets N which is typically 10^7 . This gives $r \sim 2 \mu$ for the drop radius if we take $n_0 \sim 2 \times 10^{17} \text{ cm}^{-3}$. Investigations at liquid helium temperature of the scattering of 3.39 μ light by a crystal of pure Ge excited optically by a lamp or a laser also gave good proof of the existence of EHD. These experiments show that electron-hole drops are certainly spherical and, from the angular distribution of the scattered light, one gets their size which is of the order of some microns.

A lot of other data and arguments⁶ have been put forward to support the existence of electron-hole drops, and it is generally admitted now that they have been observed in Ge, Si and Ge-Si alloys. More than 250 papers have been published in this field during the last six years, so it is not possible to give here an exhaustive list of all the

investigations which have been done to understand this condensed phase. However it may be interesting to note that, at least in Ge, many of the most important EHD parameters have been measured. This is, for example, the case of the following ones: density of electron-hole pairs in the liquid phase⁶ ($n_0 \sim 2 \times 10^{17} \text{ cm}^{-3}$), EHD binding energy⁶ with respect to gaseous excitons ($\sim 2 \text{ meV}$), EHD lifetime⁶ (40 μ sec), electric charge at the drop surface⁷, surface energy⁸ of drops ($\sim 2 \times 10^{-4} \text{ ergs/cm}^2$), critical temperature⁹ (6.5 K). It should be pointed out that theory¹⁰ and experiment are generally in very good agreement.

On the other hand, some interesting aspects of EHD are not yet well known or understood. Among others one can quote the following ones: the magnetic properties of drops have not yet been investigated; it seems that EHD can move rather rapidly⁶, but this is not explained; the nucleation process (or processes) involved in drop formation is still unclear and raises very interesting and stimulating problems whose understanding might be helpful in other areas. One could also wonder whether this liquid phase may give rise, under conditions which are to be defined, to superfluidity or superconductivity. This may seem highly speculative, but the answer to this question is not clear.

To conclude, the existence of electron-hole drops in semiconductors is now well established. This liquid phase, which presents many interesting features, results from the condensation of gaseous excitons, and this phenomenon corresponds to a first order phase transition. Many aspects of EHD are now understood, but a number of challenging experimental and theoretical problems remain. Let

us note that this condensed phase is probably the simplest example of liquid metal which is available at the present time. As such it provides a good opportunity to test theoretical methods of great interest in many body problems. Finally, it seems that this field remains very attractive for many solid state physicists and is in fact still developing.

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The Chemical Physics Section of the Atomic Physics Division

The recently-constituted Chemical Physics Section of the Atomic Physics Division will be holding its first conference from the 30th of August to the 1st of September 1976 in Amsterdam. The theme is "Exploring the chemical bond. New developments".

People interested in attending the meeting should contact Dr. R. Block, Congress Bureau (IBD) A-76, Vrije Universiteit, de Boelelaan, 1105, Amsterdam, The Netherlands or the Conference Secretary, Miss A.-M. Bonino, Batelle Institute, Advanced Studies Center, 7, route de Drize, CH-1227 Carouge-Geneva, Switzerland.

On this occasion the Section will hold its first General Meeting and the Provisional Board has taken this opportunity to announce the creation of the Section to the various European Chemical Societies. Anyone interested in participating in the activities of the Section should contact either the Chairman of the Provisional Board Professor E.A.C. Lucken, Département de Chimie Physique, Sciences II, 30, Quai E. Ansermet, 1211 Geneva 4, Switzerland or the Secretary, Professor G. Wagnière, Département de Chimie Physique, Université de Zurich, Zurich, Switzerland.

The Provisional Board has decided to couple each of its meetings with a scientific lecture. It is particularly appropriate that the first of these will be given by Professor W. Kolos who, together with Professor L. Jansen, was largely responsible for the setting up of this Section. The title of the talk, to be given in the Physical Chemistry Department of the University of Zurich on the 16th of January 76 will be "Ab Initio Calculations on the Excited States of the H₂ Molecule".

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